

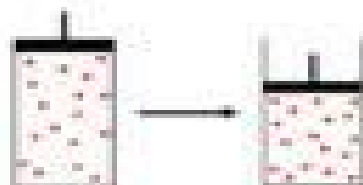
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Gas Law Practice

For the mathematical questions, show all of your work. For the conceptual questions, provide a thorough explanation for your answers.

[1] Convert 110°C into Kelvin.

A gas in a fixed cylinder as shown below is compressed while at a constant temperature. Initially, the pressure was exactly 1.00 atm, at a volume of 2.00 L. What is the final pressure when the cylinder is compressed to 1.40 L?



[2] A mass of neon occupies 500 mL at 100°C . Find its volume at 25°C , with the pressure remaining constant.

[3] A steel tank contains carbon dioxide at 27°C at a pressure of 6.0 atm. Determine the internal gas pressure when the tank and its contents are heated to 100°C .

Dalton Gas Law Practice Problems Key

**Christopher R. Hren, John T.
Moore, Peter J. Mikulecky**



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