# Compounds and Their Bonds

9

#### A. Jone: Transfer of Electrons.

Clement	Attended	I. Electron Configuration of Atom	2. Electron- Des Symbol	A. Care. or Guin	K. Electron Configuration of Son	Charge		of Non
Sodon	10.	31/21/21/21	907	Seed Let	1/2/201	930	24.0	Northern
Nigrigon	7	14/24/267	- 84 1	Georgia	14/24/29*		Nº .	Principle for
Alamanan	13	14/24/29/34/39/	- 701	Less har	14/24/29*		W.	Albaniman inte
CMonree:	12	14/24/29/34/39/	i cit	Com Let	16/26/29/36/39*		CI*	Chlorate in
Culcium	266	16/26/2g/16/2g/86/	Ca.	Loss 24	16/26/29/29/29/2	- #	Kar.	Colstant to
Otogen	- 8	14/24/2p*	100	Coin 2'er	14/24/29*	100	0.11	Orode me

#### B. Jonic Compounds and Formulas

#### 1. Physical peoperties

Compound	Appearance	Donaity	Midding Point
Sodium chloride, NaCl	White, solid, crystalline	2.36 g/cm²	806°C

#### 2. Formulias of lonic compounds

Name	Positive less	Negative loss	Forms with
Lithium indide	13.5	15	1.11
Alumimaen oxide -	ALP	0.5	:AlaOs
Calcium sulfide	Ca 3º	8.1	CoS
Magnesium brosoide	Nig 31	Br 1	MgBr
Potassium nitride	K*	Nº 5	KN
Sodium (hopeide	No. *	P ·	NaF

#### A. Names of looks compounds

Formula.	Positive for	Negative las	Name
K 38	K*	8.5	Potassium sulfide
Blaf2	Ba *	F *-	Barium fluoride
MgO	Mg 2"	0.5	Magnesium oxide
Na <sub>3</sub> N	No. 11	8.5	Sodium mitride
ASCI)	Al-*	CI.	Aluminum chłoride
MggPg	Mg **	pa-	Magmesium phosphide

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