



CCNP SWITCH 6.0

Instructor Lab Manual

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Cisco Ccnp Instructor Lab Manual

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delivered courses for the Information and Telecommunication Systems for Business and the Police Executive Leadership Graduate Programs at John Hopkins University and provided consulting and training for numerous local businesses throughout her career

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web we utilize the equation that relate amount remaining initial mass and number of half lives $n = \frac{t}{T} \times n_0$ 1 2 4 x 50 3 125 g 5 what is the half life of an isotope that is 80 remained after 16 days solution remaining 80 100 therefore $n = \frac{t}{T} \times n_0$ 100 now using the half life equation in example 1 we have

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web solution if we compare the time that has passed to the isotope s half life we note that 44 0 s is exactly 4 half lives so using the previous equation n 4 substituting and solving results in the following less than one third of a gram of 20 f remains skill building exercise the half life of 44 ti is 60 0 y

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web solution 7 2 2 4 3 half lives 1 2 0 125 the amount remaining after 3 half lives 100 0 g x 0 125 12 5 g remaining problem 2 pd 100 has a half life of 3 6 days if one had 6 02 x 10 atoms at the start how many atoms would be present after 20 0 days solution 20 0 3 6 5 56 half lives

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