

# Covalent Bonding

## How does a covalent bond form?

- A **covalent bond** forms when two atoms **share** one or more pairs of electrons.
- Atoms in covalent bonds typically have **similar electronegativities** and are **usually non-metals**.
- The shared electrons allow each atom to achieve a full outer electron shell, similar to the noble gas configuration.

## Types of Covalent Bonds

- **Single Bond:** One pair of electrons is shared (e.g.,  $\text{H}_2$ ,  $\text{Cl}_2$ ).
- **Double Bond:** Two pairs of electrons are shared (e.g.,  $\text{O}_2$ ,  $\text{CO}_2$ ).
- **Triple Bond:** Three pairs of electrons are shared (e.g.,  $\text{N}_2$ ).



## Dot-and-Cross Diagrams

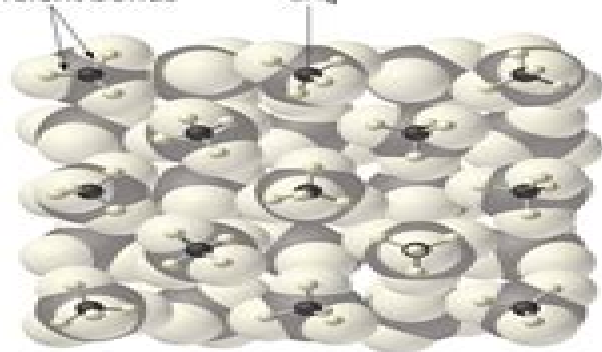
- Only the outer electron shells (valence electrons) are shown.
- Shared pairs of electrons between atoms are shown as pairs of dots and crosses.
- It represents how the atoms bond and achieve stable electron configurations.

## Example: Water $\text{H}_2\text{O}$



Covalent bonds

$\text{CH}_4$



## Properties of Covalent Compounds

- **Molecular Structure:** They are made up of discrete molecules with **weak intermolecular forces**.
- **Melting and Boiling Points:** Generally **lower** than ionic compounds because the forces between molecules are weaker.
- **Electrical Conductivity:** **Poor conductors** of electricity in all states, as they do not have free ions or electrons.
- **Solubility:** Often **soluble in nonpolar solvents** and may be insoluble in water, depending on their polarity.

# Chemistry Study Guide Covalent Bonding

**Huangqi Zhang**



## **Chemistry Study Guide Covalent Bonding:**

## Decoding **Chemistry Study Guide Covalent Bonding**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Chemistry Study Guide Covalent Bonding**," a mesmerizing literary creation penned with a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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