

Covalent Bonding

How does a covalent bond form?

- A **covalent bond** forms when two atoms **share** one or more pairs of electrons.
- Atoms in covalent bonds typically have **similar electronegativities** and are **usually non-metals**.
- The shared electrons allow each atom to achieve a full outer electron shell, similar to the noble gas configuration.

Types of Covalent Bonds

- **Single Bond:** One pair of electrons is shared (e.g., H_2 , Cl_2).
- **Double Bond:** Two pairs of electrons are shared (e.g., O_2 , CO_2).
- **Triple Bond:** Three pairs of electrons are shared (e.g., N_2).



Dot-and-Cross Diagrams

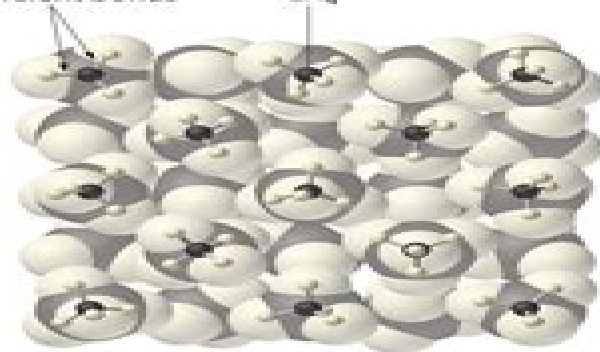
- Only the outer electron shells (valence electrons) are shown.
- Shared pairs of electrons between atoms are shown as pairs of dots and crosses.
- It represents how the atoms bond and achieve stable electron configurations.

Example: Water H_2O



Covalent bonds

CH_4



Properties of Covalent Compounds

- **Molecular Structure:** They are made up of discrete molecules with **weak intermolecular forces**.
- **Melting and Boiling Points:** Generally **lower** than ionic compounds because the forces between molecules are weaker.
- **Electrical Conductivity:** **Poor conductors** of electricity in all states, as they do not have free ions or electrons.
- **Solubility:** Often **soluble in nonpolar solvents** and may be insoluble in water, depending on their polarity.

Chemistry Study Guide Covalent Bonding

Victor M. Corman



Chemistry Study Guide Covalent Bonding:

Unveiling the Power of Verbal Artistry: An Mental Sojourn through **Chemistry Study Guide Covalent Bonding**

In a world inundated with monitors and the cacophony of immediate communication, the profound power and emotional resonance of verbal beauty frequently fade in to obscurity, eclipsed by the continuous assault of noise and distractions. However, nestled within the lyrical pages of **Chemistry Study Guide Covalent Bonding**, a fascinating work of literary elegance that pulses with raw thoughts, lies an unique trip waiting to be embarked upon. Penned by way of a virtuoso wordsmith, this magical opus instructions visitors on a psychological odyssey, softly exposing the latent potential and profound impact embedded within the complex internet of language. Within the heart-wrenching expanse of the evocative analysis, we will embark upon an introspective exploration of the book is main subjects, dissect its fascinating writing fashion, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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