

Unit One Chemistry Exam

Multiple Choice 20 marks

You need to record your answers in the appropriate section in the ANSWER BOOKLET.

Question 1

A nanoparticle is best described as

- A. one atom.
- B. one thousand atoms.
- C. one mole of atoms.
- D. all of the above are correct.

Question 2

An element in group 2 and period 3 of the periodic table could have an electron configuration of

- A. $1s^2 2s^2 2p^6 3s^2 3p^2$
- B. $1s^2 2s^2 2p^6 3s^2$ or $1s^2 2s^2 2p^6 3s^2 3p^2$
- C. $3s^2$
- D. $1s^2 2s^2 2p^1$

Question 3

An emission spectrum is produced when the valence electrons of an atom

- A. become excited.
- B. move back down to a lower energy level.
- C. remain in the ground state.
- D. undergo a transition that brings them further away from the nucleus.

Question 4

Avogadro's number is best described as

- A. the number of atoms in one mole of carbon-12.
- B. the number of atoms present in 12.011 g of the carbon – 12 isotope.
- C. the number of grams of the carbon – 12 isotope present in one mole of carbon.
- D. the number of atoms present in 12 g of any element.

Question 5

The relative molecular mass of a compound is assigned the units

- A. kJ mol^{-1}
- B. gmol^{-1}
- C. mol
- D. none of the above

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D Keegan

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