

SAMAGRA SHIKSHA KERALA
FIRST TERM EVALUATION 2023-24

Std: IX

CHEMISTRY

Score : 40

Time : 1 ½ Hours

Instructions

- First 15 minutes is given as cool off time. This time is to be spent for reading and understanding the questions.
- Answer the questions according to the directions.
- Score and time are to be considered while answering.

Answer any 4 questions from 1 to 5. Each carries 1 Score.

(4 × 1 = 4)

1. Maximum number of electrons that can be accommodated in the outermost shell of an element is ----- (1)
(12, 10, 8, 18)
2. The smallest particle of an element that can take part in chemical reactions is called ----- (1)
3. Fill up suitably. (1)
Electron : J.J Thomson
Neutron : -----
4. Complete the chemical equation. (1)
 $\text{Na} \rightarrow \text{---} + \text{e}^-$
5. The elements belonging to group 18 in the periodic table are called.----- (1)

Answer any 4 questions from 6 to 10. Each carries 2 Scores.

(4 × 2 = 8)

6. Symbols of some atoms are given. (symbols are not real)



- a) Which among these is an isotopic pair? Give reason. (1)
- b) How many neutrons are present in Q? (1)
7. Analyse the chemical equation given below. (1)
 $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$
- a) Which are the products? (1)
- b) Find the total number of atoms present in the reactant side? (1)
8. Some elements and their valencies are given.

Element	Valency
Ba	2
Cl	1
O	2

- a) Write the chemical formula of Barium chloride. (1)
- b) The chemical formula of Calcium oxide is CaO. What is the valency of Ca? (1)

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