



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Their Formulas
(Show the steps for naming the compound)



Steps for Naming:

1. Identify the cation (metal) and anion (non-metal).
2. Name the cation first, followed by the anion.
3. Use the appropriate suffix for the anion (usually -ide).
4. If the cation is a transition metal, include its oxidation state in parentheses.
5. If the anion is a polyatomic ion, use its name.
6. If the compound is a binary compound, use the prefix system.

Binary Compounds

Formula	Name
NaCl	Sodium Chloride
CaO	Calcium Oxide
Fe ₂ O ₃	Iron(III) Oxide
Al ₂ Cl ₆	Aluminum Chloride
CO ₂	Carbon Dioxide
SO ₂	Sulfur Dioxide
NO ₂	Nitrogen Dioxide
CH ₄	Methane
C ₂ H ₆	Ethane
C ₃ H ₈	Propane
C ₄ H ₁₀	Butane
C ₅ H ₁₂	Pentane
C ₆ H ₁₄	Hexane
C ₇ H ₁₆	Heptane
C ₈ H ₁₈	Octane
C ₉ H ₂₀	Nonane
C ₁₀ H ₂₂	Decane

Common Ions and Polyatomic Ions

Formula	Name
Na ⁺	Sodium Ion
Ca ²⁺	Calcium Ion
Fe ²⁺	Iron(II) Ion
Fe ³⁺	Iron(III) Ion
Al ³⁺	Aluminum Ion
CO ₃ ²⁻	Carbonate Ion
SO ₄ ²⁻	Sulfate Ion
NO ₃ ⁻	Nitrate Ion
CH ₃ ⁺	Methyl Cation
C ₂ H ₅ ⁺	Ethyl Cation
C ₃ H ₇ ⁺	Propyl Cation
C ₄ H ₉ ⁺	Butyl Cation
C ₅ H ₁₁ ⁺	Pentyl Cation
C ₆ H ₁₃ ⁺	Hexyl Cation
C ₇ H ₁₅ ⁺	Heptyl Cation
C ₈ H ₁₇ ⁺	Octyl Cation
C ₉ H ₁₉ ⁺	Nonyl Cation
C ₁₀ H ₂₁ ⁺	Decyl Cation

Naming Compounds with a Central Atom and Anion



Carbon Dioxide

Steps for Naming:

1. Identify the central atom and the anion.
2. Name the central atom first, followed by the anion.
3. Use the appropriate suffix for the anion (usually -ide).
4. If the central atom is a transition metal, include its oxidation state in parentheses.
5. If the anion is a polyatomic ion, use its name.

Naming Practice

Formula	Name	Formula	Name
NaCl	Sodium Chloride	CaO	Calcium Oxide
Fe ₂ O ₃	Iron(III) Oxide	Al ₂ Cl ₆	Aluminum Chloride
CO ₂	Carbon Dioxide	SO ₂	Sulfur Dioxide
NO ₂	Nitrogen Dioxide	CH ₄	Methane
C ₂ H ₆	Ethane	C ₃ H ₈	Propane
C ₄ H ₁₀	Butane	C ₅ H ₁₂	Pentane
C ₆ H ₁₄	Hexane	C ₇ H ₁₆	Heptane
C ₈ H ₁₈	Octane	C ₉ H ₂₀	Nonane
C ₁₀ H ₂₂	Decane		

Common Names for Specific Compounds

Formula	Common Name	Formula	Common Name
NaCl	Salt	CaO	Quicklime
Fe ₂ O ₃	Rust	Al ₂ Cl ₆	Aluminum Chloride
CO ₂	Carbon Dioxide	SO ₂	Sulfur Dioxide
NO ₂	Nitrogen Dioxide	CH ₄	Methane
C ₂ H ₆	Ethane	C ₃ H ₈	Propane
C ₄ H ₁₀	Butane	C ₅ H ₁₂	Pentane
C ₆ H ₁₄	Hexane	C ₇ H ₁₆	Heptane
C ₈ H ₁₈	Octane	C ₉ H ₂₀	Nonane
C ₁₀ H ₂₂	Decane		

Chemistry Naming Compounds Study Guide

RJ Shavelson



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