

Let's Be Rational

**Understanding
Fraction
Operations**

Lappan, Phillips, Fey, Friel

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web mar 13 2023 because the masses of atoms are so small it is more convenient to give nuclide masses in atomic mass units abbreviated amu or u the latter is the official si abbreviation rather grams the atomic mass unit is defined as follows one atomic mass unit is defined as 1 12 of the mass of ce 12 6 c atom

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web if you start with 14 82 g of ca oh 2 c a o h 2 and 16 35 g of h2so4 h 2 s o 4 a determine the limiting reagent b determine the number of moles of h2o h 2 o produced c determine the number of grams of caso4 c a s o 4 produced d determine the

number of grams of excess reagent left 1 make sure the equation is balanced

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